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Preparation and structural analysis of ordered vacancy compounds in

the Cu-In-Se system

S.Nomura and S.Endo

Department of Electrical Engineering, Faculty of Engineering, Science University of Tokyo 1-3 Kagurazaka, Shinjuku-ku, Tokyo 162

The possibility of forming the OVC structure has been investigated for a variety of compositions not only on the Cu₂Se-In₂Se₃ binary tie line but also in the vicinity of the tie line. The authors attempted to obtain OVC crystals both by quenching and by normal freezing method. Each sample crystal phase and the homogeneity are checked using the x-ray diffraction measurement and the electron probe microanalysis. These experimental results reveal that the OVC phase allows a variety of compositions around the starting one such as CuIn₂Se_{3.5}. This can be explained by assuming Cu+In+ \Box site for the ordered vacancy formation.

1. INTRODUCTION

It has been found that the ordered vacancy chalcopyrite (OVC) compound, CuIn₃Se₅, forms a surface layer of indiumrich CuInSe₂ thin films [1]. This suggests that the OVC phase would play an important role in the homo and/or hetero-junction devices including CuInSe₂ layer. However, little has been known about the fundamental properties, because single crystals of the OVC phases have not been successfully obtained except for some thin-film synthesis [2]. Only two phases, CuIn₃Se₅ and CuIn₂Se_{3.5}, are known as the possible OVC phases, both of which exist on the Cu₂Se-In₂Se₃ binary tie line.

Hönle et al [3] gave the detailed crystal structure for one example of the OVC (β) phase, which was informative for a possible atomic arrangement of the OVC phase. However, the OVC structure was observed to be flexible in composition, showing various compositions even out of the binary tie line in our experimental results. There has not been detailed discussion on the mechanism allowing the compositional variety.

In this work, the possible OVC structures allowing the compositional variation are proposed in view of the simulation of the x-ray diffraction intensity in comparison with the experimental results.

2. EXPERIMENTAL

2.1 Sample Preparation

We attempted to synthesize two possible OVC phases of CuIn₃Se₅ and CuIn₂Se_{3.5}. Compositional elements of the respective 6N purity were used for the sample preparation. These elements were sealed in a quartz tube (16 $\phi \times 120$ mm) under a vacuum of $\sim 10^{-6}$ Torr ($\sim 10^{-4}$ Pa), heated up to 1100°C, using an electric furnace, held for 8 hours, and then gradually (0.1deg/min) cooled down to the temperatures forming the respective OVC phases. After holding the temperature for 3-5 days, the last step of crystallization was performed both with quenching from the temperatures (normally 700-900 °C) and with gradual cooling (0.1deg/min) down to room temperature.

Table 1. Atomic positions (x, y) of A, B, and			
C sublattices of the tetragonal chalcopyrite			
structure, u=0.224 for CuInSe ₂ [5].			

z (n/8)	site	site
7	C7(1/4, 1/2-u)	$C_8(3/4, 1/2+u)$
6	A4 (0, 1/2)	B4 (1/2, 0)
5	C5(1-u, 1/4)	$C_6(u, 3/4)$
4	$A_3(0, 0)$	$B_3(1/2, 1/2)$
3	C3(1/4, u)	$C_4(3/4, 1-u)$
2	$A_2(1/2, 0)$	$B_2(0, 1/2)$
1	$C_1(1/2+u, 1/4)$	$C_2(1/2-u, 3/4)$
0	A_1 (1/2, 1/2)	B ₁ (0, 0)

2.2 EPMA measurement and x-ray diffraction

Sample compositions were analyzed by the EPMA (Electron Probe Microscope Analysis) measurement. Powder x-ray (Cu-K α) diffraction (XRD) was used to determine crystal phases and the lattice constants and to measure diffraction intensity of each diffraction line for the comparison with the calculated results. In order to measure both the diffraction angle and the intensity precisely, the diffraction angle 2 θ was scanned at a rate of 1/32 deg/min.

8. CALCULATION OF DIFFRACTION INTENSITY

Relative x-ray diffraction intensity was calculated for the respective diffraction lines assuming possible OVC structures, using the following equation [4]:

$$I = |\mathbf{F}(\mathbf{h})|^2 \cdot \frac{1 + \cos^2 2\theta}{\sin^2 \theta \cos \theta} \cdot m \tag{1}$$

where F(h) is the crystal structure factor followed by its correction factors as a function of diffraction angle θ relating to both the polarization of an x-ray photon and the divergence of an incident x-ray, and m is the multiplicity of diffraction. The crystal structure factor F(h) can be determined by the atomic scattering factor $f_j(h)$, Miller indices h=(hkl) and the coordinates of atoms (x_j, y_j, z_j) for the j-th atom as follows:

$$\boldsymbol{F(h)} = \sum_{j} f_{j}(\boldsymbol{h}) \exp\left\{2\pi i(hx_{j} + ky_{j} + lz_{j})\right\} \quad (2)$$

Atomic coordinates of the tetragonal chalcopyrite structure are listed in Table 1.

4. Results and Discussion

4.1 Phase formation

Homogeneous single phase was easily obtained from the starting composition of Cu:In:Se=1:2:3.5 even on the crystallization with slow cooling rate (0.1deg/min). On the other hand, homogeneous synthesis was not attained from the starting composition of Cu:In:Se=1:3:5 even by quenching at temperatures (700-900°C) expected to form the OVC phase [6]. The resultant crystals were inevitably followed by a copperdeficient layer phase. However, it was easy to distinguish the layer phase material from OVC crystals. Figure 1 shows the x-ray diffraction profiles of (a) CuIn₂Se_{3.5} and (b) CuIn₃Se₅ crystals. Although these diffraction profiles are quit similar to that of CuInSe₂, small but definite three diffraction i.e. (110), (202),lines, and (114).characteristic of OVC phase are seen. The lattice constants of these two phases were determined as a=0.5756 nm, c=1.1525 nm for CuIn₂Se_{3.5} and a=0.5757 nm, c=1.1534 nm for CuIn₃Se₅. Composition rates Cu:In:Se of these powdered samples used for the XRD measurement were determined by the EPMA as 15.35:29.47:55.18 for CuIn₂Se_{3.5} and 11.28:33.69:55.03 for CuIn₃Se₅.



Figure 1. X-ray diffraction profiles of two OVC phases of (a) CuIn₂Se_{3.5} and (b) CuIn₃Se_{5.}



Figure 2. A layer copper-deficient phase found additional to CuIn₃Se₅ crystals.

Figure 2 shows a diffraction profile of the copper-deficient layer material found being segregated from CuIn₃Se₅ crystals. The phase separation was remarkable when the quenching temperature was 900°C. The average composition of this substance is Cu:In:Se = 8.57:34.50:56.93, approximately 1:4:6.5 which corresponds to Cu₂Se (20%) + In₂Se₃ (80%) on the binary tie line. Copperdeficient ternary phases were reported as ε , ζ , and η in ref.4. These phases are richer in In₂Se₃ composition than the layer phase found in our samples.

4.2 OVC structure model

In the case of CuInSe₂, Three sublattices, A, B, and C-site listed in Table 1 are fully occupied by Cu, In, and Se atoms, respectively. The unit cell has $4ABC_2$ atoms. A general expression of composition, Cu_xIn_ySe_z, can be equally replaced in the form of A₄B₄C₈ with:Cu_{8x/z}In_{8y/z} $\Box_{8[1-(x+y)/z]}Se_8$, where the symbol \Box expresses vacancies. Vacancies in anion (Se) site have not been considered in this work. The atomic portions of Cu and In atoms are Cu:In: \Box =2:4:1 for CuIn₂Se_{3.5} and 1:3:1 for CuIn₃Se₅.

Figure 3 shows the relationship between Cu/Se and In/Se molar ratios detected by the EPMA spot data for the crystal growth surface of OVC samples. The data points distribute around Cu₂Se-In₂Se₃ binary tie line. The expected OVC region is 1/2 < x/z + y/z < 1, which is shown by dotted lines in Fig.3. The range of distribution is wider along the In/Se axis. The respective data distributions of CuIn₂Se_{3.5} and CuIn₃Se₅ clearly share the neighboring regions along the Cu/Se axis.

These results can be explained by assuming the flexible atomic occupation of cation site. Assuming that Cu/Se is nearly constant for the respective CuIn₂Se_{3.5} and CuIn₃Se₅ phases, the portion of vacancies vhas the following relation with In/Cu ratio, i.e. y/x,

$$y = 8 \left[1 - Const. (1 + y / x) \right].$$
 (3)

The constant parameter is 2/7 for CuIn₂Se_{3.5}, and 1/5 for CuIn₃Se₅. The OVC structures are flexible for In/Cu ratio with changing the portion of vacancies v. This suggests that vacancies unlikely exist as independently unoccupied sites, but exist such as Cu+ \Box , In+ \Box , or Cu+In+ \Box site.



Figure 3. Relationship between Cu/Se and In/Se molar ratios. Symbols denote: \bigcirc q-CuIn₂Se_{3.5}, \bigcirc q-CuIn₃Se₅, \blacksquare nf-CuIn₂Se_{3.5}, \square nf-CuIn₃Se₅ (q:quenched, nf:normal freezing).



Figure 4. Diffraction profiles of (a) $CuIn_2Se_{3.5}$ and (b) $CuIn_3Se_5$ simulated according to the site occupation listed in Table 2.

Taking tetragonal symmetry into account, each site occupation is proposed as listed in Table 2. Difference between the two OVC phases is that atomic sites fully occupied by Cu exist in CuIn₂Se_{3.5}. Figure 4 shows the calculated diffraction patterns for CuIn₂Se_{3.5} and CuIn₃Se₅ assuming the site occupation listed in Table 2. The results of simulation successfully reproduce the experimental diffraction profiles.

Table 2

Proposed site occupation for the two OVC phases.

Site	CuIn ₂ Se _{3.5}	CuIn3Se5
A3, B1	Cu	Cu _{0.8}
A2, B4	In	In
A1, A4,	Cu0.071In0.643 0.286	$In_{0.7}\square_{0.3}$
B_2, B_3		

5. Summary

Two OVC phases in the Cu-In-Se system, CuIn₂Se_{3.5} and CuIn₃Se₅, were investigated. Crystal structures allowing the compositional variety were proposed using the simulation of x-ray diffraction in comparison with the experimental results, assuming Cu+In+ \Box site for ordered vacancy formation.

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