Electronic and Magnetic Properties of Perovskite Manganites Including Rare Earth Metal Ions

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Electronic and magnetic properties of perovskite manganites of $(La_{1-x}R_x)_{0.7}A_{0.3}MnO_3$ (R=Gd, Dy, Ho, Er; A=Ca, Sr) have been studied by measuring electrical resistivity and magnetic susceptibility in order to know the relations between their electronic and magnetic properties and two parameters; tolerance factor, t (geometrical index), and the variance, σ^2 (the second moment), of A-site ion radii distribution. We found: (i) Curie temperature, T_c , where metal-insulator transition occurs, decreases with decreasing t. (ii) At the constant t, T_c also decreases with increasing σ^2 . (iii) when t is smaller than 0.908 and σ^2 is larger than $2.0 \times 10^{-5} nm^2$, the compounds show spin-glass insulator state below T_g . Considering these results, we conclude that electronic and magnetic properties of manganites depend not only on t but also on σ^2 .

Key words: perovskite, rare earth metal ions, metal-insulator transition, Curie temperature, tolerance factor

1. INTRODUCTION

Perovskite manganites with 20-40% of trivalent rare-earth ions of R^{3+} replaced by divalent alkaline-earth metal ions of A^{2+} , $R_{1-x}A_xMnO_3$, are found to show remarkable phenomena such as metalinsulator transition and colossal magnetoresistance (CMR) and are expected to be applied to electronic devices. Therefore, perovskite manganites have been extensively studied from the 1950s. According to the studies¹⁻³, electronic and magnetic properties of those manganites are mainly explained by double-exchange interaction, ferromagnetic interaction between Mn^{3+} and Mn^{4+} that is caused by the hopping of e_g electrons between the two partially filled *d* shells with strong on-site Hund's coupling⁴.

Hwang et al.⁵ have investigated (La₁. ${}_{x}R_{x})_{0.7}Ca_{0.3}MnO_3$ (R=Pr³⁺, Y³⁺) systematically while keeping the Mn³⁺/Mn⁴⁺ ratio fixed at 7/3. They found that T_{C} decreases with increasing the content of Pr³⁺ and/or Y³⁺. Considering these results, they concluded that the electronic and magnetic states of these compounds depend on tolerance factor, t. Incidentally the tolerance factor, a geometrical index, is defined as $t=(r_A+r_O)/\sqrt{2} (r_{Mn}+r_O)$, where r_A is the average ionic radius of \mathbb{R}^{3+} and \mathbb{Ca}^{2+} , r_O is the ionic radius of \mathbb{O}^{2-} , and r_{Mn} is the average ionic radius of \mathbb{Mn}^{3+} and \mathbb{Mn}^{4+} .

However, recently, Rodriguez-Martinez and Attfield⁶ speculated that the difference in size among A-site ions also influences electronic and magnetic properties of manganites. In order to quantify such a difference, they introduced the variance, σ^2 (the second moment), of the A-site ions radii distribution. Incidentally, σ^2 is defined as $\sigma^2 = \sum y_i r_i^2 - \langle r_i \rangle^2$, where i is difference of the kinds of A-site ions, r_i is the radius of each A-site ion, y_i is its fractional occupancy ($\sum y_i=1$), and $\langle r_i \rangle$ is the average radius of A-site ions. In the present study, therefore, we have investigated the relation between these parameters (t, σ^2) and electronic and magnetic properties of, $(La_{1-x}R_x)_{0.7}A_{0.3}MnO_3$ (R=Gd, Dy, Ho, Er; A=Ca, Sr).

2. EXPERIMENTAL

 $(La_{1-x}R_x)_{0.7}A_{0.3}MnO_3$ compounds were prepared by conventional solid state reaction processing in air (calcined at 1273K for 12h, sintered at 1623K for 48h). *t* was determined by using ionic radii reported by Shannon⁷. Powder X-ray diffraction measurement was made at room temperature. Resistivity and magnetic susceptibility were measured by the standard four-probe method and by vibrating sample magnetometer (VSM), respectively.

3. RESULTS

3.1 X-ray diffraction measurement

From X-ray measurement, most compounds used show a single phase with a cubic structure at room temperature. These lattice constants change from 0.773 nm (t=0.916) to 0.763 nm (t=0.898) with decreasing t.

3.2 Resistivity measurements

We made resistivity measurements for the compounds with the cubic structure. Typical results of (La_{1-x}Dy_x)_{0.7}Ca_{0.3}MnO₃ are shown in Fig. 1, where metal-insulator transition temperature, $T_{\rm MI}$, indicated with a single arrow, is observed. As known from the figure, $T_{\rm MI}$ decreases with decreasing t and goes to below 1.5K for t=0.898. It should be noted in the figure that resistivities of t=0.911, 0.908 and 0.907 increase again below temperatures indicated with double arrows, which correspond to the temperature of a spin-glass transition, which will be described later. Similar behavior of the electrical resistivity as a function of temperature are observed in the other compounds replaced by the other rare earth metal ions (Gd^{3+}) , $Ho^{3+}, Er^{3+}).$



Fig.1 Temperature dependence of resistivity for $(La_{1,x}Dy_x)_{0,7}Ca_{0,3}MnO_3$.

3.3 Magnetic susceptibility measurements

We made magnetic susceptibility measurements on the two conditions; one is the zero-field-cooled (ZFC) run and the other is field-cooled (FC) run at H=0.01T. A typical result of the compounds including Dy³⁺ ions is shown in Fig. 2, where solid and dotted curves represent the results of ZFC and FC runs, respectively.



As known from the figure, Curie temperature, $T_{\rm C}$, indicated with a single arrow, decreases with decreasing t. It should be noted that M/H-T curves of t=0.911, 0.908, 0.907 and 0.898 obtained by ZFC runs are different from those obtained by FC runs below temperatures indicated with double arrows. Similar temperature dependence of M/H-T curves were observed for (La₁₋ $_{x}Tb_{x})_{2/3}Ca_{1/3}MnO_{3}^{8}$ and Fe-Au⁹ and they were confirmed to show spin-glass states by neutron diffraction measurements. Therefore, the different behavior between ZFC and FC runs in the present compounds is believed to be due to a spinglass transition. According to the previous studies^{8,9}, the transition temperature, $T_{\rm g}$, is defined as a temperature where M/H-T curve of ZFC run yields to a cusp, in cases that $T_{\rm C}$ and $T_{\rm g}$ are close each other or identical. This situation holds for the compound of t = 0.908, 0.907 and 0.898 and, therefore, T_{g} , is determined as a temperature of the observed cusp, which is shown in each M/H-T curve with the double arrow. On the other hand, for the compound of t=0.911, M/H-T curve under a ZFC run has two cusps, being quite

similar to that of a compound that becomes a typical reentrant spin-glass at low temperatures⁹. In this case, T_g is determined as a temperature where the second cusp appears (indicated with a double arrow in the figure).

4. DISCUSSION

Figure 3 shows $T_{\rm C}$ and $T_{\rm g}$ of the present $({\rm La}_{1-{\rm x}}{\rm R}_{{\rm x}})_{0.7}{\rm Ca}_{0.3}{\rm MnO}_3$ as a function of t. The figure also show $T_{\rm C}$ of the previous $({\rm La}_{1-{\rm x}}{\rm Pr}_{{\rm x}})_{0.7}{\rm Ca}_{0.3}{\rm MnO}_3$ obtained by Hwang et al.⁵ In the figure, electronic and magnetic states in the present compounds are also shown by using notations derived previously^{5,8}; that is, paramagnetic insulator (PMI), ferromagnetic metal (FMM), and spin-glass insulator (SGI).



Fig.3 Electronic and magnetic phase diagram of $(La-R)_{0,7}Ca_{0,3}MnO_3$.

It is noted from the figure, when t is larger than 0.913, $T_{\rm C}$ decreases with decreasing t, being similar to the previous study by Hwang et al.⁵ However, when t takes values between 0.913 and 0.907, FMM state and reentrant spin-glass insulator state appears below $T_{\rm C}$ and $T_{\rm g}$, respectively and when t is smaller than 0.907, SGI state appears, being different from the previous study by Hwang et al.⁵ It should be also noted that $T_{\rm C}$ of present compounds are smaller than those of previous compounds although t takes the same value. These results suggest that electronic and magnetic states of these compounds does not depend on t only. Then, we plot the relation between $T_{\rm C}$ and σ^2 in order to know the influence of σ^2 on electronic and magnetic properties of manganites, as suggested by Lodriguez-Martinez and Attfield.



The relations between $\sigma^2 - T_C$ or T_g is shown in Fig. 4, where $T_{\rm C}$ decreases with increasing σ^2 even when t takes the same value. It is also noted that when σ^2 is larger than about 2.0×10^{-5} nm² and t is larger than 0.908, the ground state changes to SGI state, meaning that σ^2 also influences the electronic and magnetic properties of manganites. Such influences of t and σ^2 on electronic and magnetic properties of manganites are speculated as follows. That is, with decreasing t, Mn-O-Mn bond angles decreases; i.e., transfer integer b decreases⁴. This causes the decrease of $T_{\rm C}$. On the other hand, with increasing σ^2 , the randomness of Mn-O-Mn bond angles distribution increases; i.e., the compounds are inhomogeneous magnetically. This causes the SGI state consisting ferromagnetic and antiferromagnetic region. Thus, we conclude that electronic and magnetic properties of manganites depend not only on t but also on σ^2 .

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