Preparation of Highly Oriented TiO₂/ZnO Films by Pulsed Laser Deposition

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Highly oriented TiO_2 films consisting of anatase TiO_2 (001) and rutile TiO_2 (100) were grown using a highly oriented ZnO (0001) buffer layer with nanometer sizes grain on quartz glass substrates. TiO_2 thin films were deposited by pulsed laser deposition (KrF excimer laser) in an oxygen atmosphere. The quality of films and crystallographic relationships were assessed by x-ray diffraction, x-ray pole figures and Rutherford backscattering spectroscopy (RBS)/channeling. The typical TiO_2 films were deposited at 500°C in 13.3 Pa of oxygen pressure. The anatase-rutile ratio in such phase mixed films was controlled by oxygen pressure during the deposition. From scanning electron microscopy observation, the deposited film was found to have a rough surface consisting of coarse grains with about 100 nm sizes. It was possible to improve photocatalytic activity for the decomposition of the organic dyes by the growth of TiO_2 on highly oriented ZnO buffer layer. Key words: Photocatalysis, TiO_2 , ZnO, PLD, XRD

1. INTRODUCTION

The photocatalytic properties of titanium dioxide (TiO_2) have attracted much interest from the viewpoints of basic science and applications. The excellent dielectric features in electronic, optical and photo-electrochemical properties of TiO₂ have been utilized for dielectric layers in thin film capacitors, anti-reflection coating layers, sensor and photocatalysts. Especially gas interesting is the photocatalytic property [1], and it can be utilized to decompose toxic gases into safe gaseous compounds using solar energy [2]. For improving photocatalytic activity of TiO₂ films, the control of crystal quality and the fabrication of effective large surface area were required. TiO₂ is known to crystallize in three different crystallographic structures: rutile (tetragonal), anatase (tetragonal) and brookite (orthorhombic), and the rutile, which is most stable, has been extensively studied and widely used. Several works were reported about epitaxial rutile and anatase films by metal-organic chemical vapor deposition (MOCVD)[3], ion beam sputtering [4, 5], pulsed laser deposition (PLD) [6-9]. However, the growth of epitaxial TiO₂ films was necessary to use the high quality single-crystal substrates such as SrTiO₃, LaAlO₃, α -Al₂O₃. In order to utilize the photocatalytic activity of epitaxial TiO₂ films, it is essential that techniques for preparation films on glass or polycrystalline substrate be available. Zinc oxide (ZnO) is one of the most suitable materials to use as a buffer layer. Since ZnO films are highly caxis oriented, self-textured ZnO thin films have been prepared on any substrate such as quartz glass [10, 11]. Furthermore, high quality epitaxial ZnO films have been synthesized on α -Al₂O₃ substrates [12-14]. The surface morphology of highly oriented ZnO films, hexagonal shaped grains with nanometer sizes have been observed [11]. Thus, ZnO buffer layer on substrate can be realized the fabrication of TiO₂ films consisting of highly oriented grains with nanometer sizes. In the present study, highly oriented TiO₂ films were grown using a highly oriented ZnO (0001) buffer layer on quartz glass substrates by pulsed laser deposition in an oxygen atmosphere. The crystal crystallographic quality, the orientation relationships between film and substrate, and photocatalytic activity of films are reported.



Fig. 1. Schematic figure of the pulsed laser deposition apparatus.

2. EXPERIMENTAL

TiO₂ and ZnO films were deposited on α -Al₂O₃ and quartz glass substrates by PLD using a KrF excimer laser (wavelength: 248 nm, Lambda Physik). Figure 1 shows the schematic figure of the PLD apparatus. The laser beam was incident on a target with an incident angle of 45°. It was focused to a 2×3 mm rectangle on the target face. Typical laser fluence and repetition rate were 150 mJ/cm² and 10 Hz. Both the target stage and substrate holder were rotated during the deposition. A metal Zn (purity: 99.9%) and a single crystal TiO₂ (rutile) targets were used for ZnO and TiO₂ film deposition, respectively. The substrates were separated by about 5 cm from the target. The deposition chamber was evacuated up to a base pressure of about 6×10^{-4} Pa using a turbomolecular pump (TMP). Oxygen gas (purity: 99.99 %) was flowed into the chamber through a mass-flow meter controlled by an absolute pressure gauge (Baratron 626, MKS) under the pumping condition. All of the substrates were obtained commercially. They are mirror-polished at both sides, and the typical size is $10 \times 10 \times 0.5$ mm³. In this study, ~ 200 nm thick TiO₂ films were deposited at 500°C under oxygen pressure of 1.3 – 26.3 Pa.

The crystallographic relationships between TiO₂ films and ZnO layer were determined by Xray diffraction measurements using a highresolution diffractometer (X'Pert-MRD, Philips) The X-ray source was operated at 40 kV and 30 mA Cu-Ka for radiations. Rutherford backscattering spectroscopy (RBS)/channeling analysis using a 3 MV single-stage-accelerator at JAERI/Takasaki was employed to characterize the epitaxial thin films. The analyzing 2.0 MeV ⁴He⁺ ions were incident and backscattered particles were detected at 165° scattering angle with a surface barrier detector. The beam size was about 1 mm in diameter and the beam current was about 10 nA typically. Samples were mounted on a 3-



Fig. 2. X-ray diffraction patterns for ZnO films on (a) α -Al₂O₃ (0001), (b) α -Al₂O₃ (11 $\overline{2}0$), and (c) quartz glass, respectively. The films were deposited at 500°C under the oxygen pressure of 4.7 Pa.

axis goniometer to pattern the planar channeling and the axial channeling in the angular coordinate. The film thickness and composition were evaluated from RBS spectra. The surface morphology of the deposited films was examined using a high-resolution field emission scanning electron microscopy (JSM6700F, JEOL). Photocatalytic activity of films was evaluated using the decomposition of methylene blue in aqueous solution monitored by UV-vis spectroscopy.

3. RESULTS AND DISCUSSION

The growth condition for highly oriented ZnO films was optimized referring to the crystal quality of ZnO film on the α -Al₂O₃ (0001) substrate. The suitable condition for highly oriented ZnO films was following conditions; laser energy: 150 mJ/cm², oxygen gas pressures: 4.7 Pa, substrate temperature: 500°C, target species: Zn metal. Figure 1 shows the X-ray diffraction patterns for ZnO film on the (a) α - Al_2O_3 (0001), (b) α - Al_2O_3 (1120), and (c) quartz glass substrates, respectively. It can be seen from Fig. 1, only the (0001) reflections from the ZnO are observed without any reflection from the substrates, which indicates that the ZnO films are grown with c-axis orientation. The in-plane orientations of ZnO films were examined through the pole figure measurement. The epitaxial relationships between ZnO and α -Al₂O₃ substrate were found to be $ZnO(0001)/\alpha$ -Al₂O₃ (0001) with



Fig. 3. 2.0 MeV ⁴He⁺ RBS/channeling spectra for deposited ZnO films on the (a) α -Al₂O₃ (11 $\overline{2}$ 0), and (b) quartz glass, respectively. The thicknesses of ZnO fims are 233 nm and 571 nm, respectively. The aligned spectrum was taken with the beam directed along the <0001> axis of the ZnO films.

ZnO [10]] $||\alpha-Al_2O_3|$ [1]20] and ZnO(0001)/ α -Al_2O_3 (1120) with ZnO [1120] $||\alpha-Al_2O_3|$ [0001], respectively. The pole figure analysis indicates that the ZnO(0001) on the α -Al_2O_3 (0001) has a twinned structure rotated by 30° with each other around the <0001> direction of the ZnO film. However the ZnO(0001) on the α -Al_2O_3 (1120) has no rotational domains. On the other hand, ZnO on quartz glass substrate has a (0001) preferred orientation with randomly oriented inplane direction. The observation of surface morphology by SEM indicates that the ZnO films consist of coarse grains with about 100 nm sizes (not shown).

The quality of epitaxial ZnO films was characterized through the RBS/channeling analysis. Figure 3 illustrates 2.0 MeV ⁴He⁺ RBS spectra for the ZnO film on (a) α -Al₂O₃ (1120) and (b) quartz glass substrates taken under the random and the axial channeling condition. The aligned spectrum was taken with the beam directed along the <0001> axis of the ZnO film. In Fig. 3(a) RBS spectra, one can recognize clearly the separated peaks from the ZnO film and α -Al₂O₃ substrate. The peaks at 1.56 MeV and 0.8 MeV correspond to the Zn component in the ZnO film and Al component in the α -Al₂O₃ substrate. Judging from the peak intensity, the high quality ZnO film is grown up from the interface and the interface is not mixed with each other within the depth resolution (~ 10 nm) of this technique. The minimum yield, χ_{min} value, the ratio between the random and the axially aligned yield at the fixed depth near the surface region, gives a measure to evaluate the degree of disorder in crystalline solids. The χ_{min} value in the <0001> aligned spectrum is 0.05 at the just area behind the surface peak of the Zn component in the ZnO film. which suggests that the crystal quality of the ZnO film is high enough as in a bulk single-crystal. The planar channeling analysis around the major axes of the ZnO film and the α -Al₂O₃ gives the



Fig. 4. X-ray diffraction patterns for TiO_2/ZnO films on quartz glass substrate. TiO_2 films were deposited at various oxygen pressures: (a) 1.3 Pa; (b) 13.3 Pa; (c) 26.6 Pa.

evidence that the ZnO <0001> crystallographic axis is parallel to the α -Al₂O₃ <11 $\overline{2}$ 0> axis. As can be seen from Fig.3 (b), axial channeling RBS spectra are obtained from ZnO film on quartz glass substrate even if it has amorphous structure. The yield at 1.56 MeV regions is from the Zn component of the ZnO film at the surface, and the huge reduction of this yield under the axial channeling condition suggests that the deposited film has high crystal quality compare to the interface region. It is found that ZnO films on the quartz glass substrate are highly (0001) oriented with increasing thickness. The results suggest that thicker ZnO films on glass substrate can be useful as a buffer layer for highly oriented oxide films.

The TiO₂ films were prepared under the following conditions; laser energy: 150 mJ/cm², oxygen pressures: 1.3 – 26.3 Pa, substrate temperature: 500°C, target species: single-crystal TiO₂ (rutile). Figure 4 shows the X-ray diffraction patterns for TiO₂/ZnO films on quartz glass substrates deposited at various oxygen pressures. It can be seen from Fig. 4 (a)-(c), the structure of deposited TiO₂ film on the highly oriented ZnO (0001) layer is composed of oriented anatase TiO_2 (001) and rutile TiO_2 (100). As the oxygen pressure was increased during the deposition, the component of anatase phase was increased. It is found that the anatase-rutile ratio in such phase mixed films is controlled by oxygen pressure during the deposition. The quality of oriented TiO₂/ZnO films was characterized through the RBS/channeling. But any ion channeling phenomena were not observed from deposited TiO₂ films on (0001) oriented ZnO layers, which suggests that the TiO₂ films have low crystal quality compare to the ZnO layers.

Figure 5 shows the surface morphology of the TiO_2 film on ZnO buffer layer observed by SEM. The TiO_2 film was grown at 500°C under the oxygen pressure of 13.3 Pa. The surface of the film shows coarse grains with about 100 nm sizes. The grain size of the TiO_2 films is comparable to that of ZnO films on quartz glass substrate. The



Fig. 5. SEM image of TiO_2 surface grown at 500°C under the oxygen pressure of 13.3 Pa.



Fig. 6. Decomposition of methylene blue in an aqueous solution $(1 \times 10^{-5} \text{ mol dm}^{-3})$ by the (\Box) TiO₂/glass and (•) TiO₂/ZnO/glass as a function of the UV light irradiation time.

surface morphology implies that each grain corresponds to single crystal of rutile and anatase TiO_2 structures. These grains have grown epitaxailly on the highly oriented ZnO (0001) buffer layer.

Photocatalytic activity of films was evaluated using the decomposition of methylene blue in aqueous solution. The progress of the degradation reaction was followed by monitoring the decrease in the absorbance of aqueous dye solutions as a function of the UV-light irradiation time counted from the beginning. The UV-light (1.0 mW/cm^2) with a black light lamp was used. The result of TiO₂/ZnO films deposited with different oxygen pressure showed that the TiO₂ film deposited on the ZnO buffer layer at 500°C under the oxygen pressure of 13.3 Pa, consisting of oriented anatase TiO₂ (001) and rutile TiO₂ (100) (referred to Fig. 4), had a high efficiency for the decomposition reaction. Figure 6 shows the photocatalytic behavior of the (D) TiO₂/ZnO and (•) TiO₂ films on quartz glass substrates during the degradation of methylene blue. Each TiO₂ films were deposited at 500°C under the oxygen pressure of 13.3 Pa. The TiO₂/ZnO film shows a high efficiency for the decomposition reaction. The above results reveal that the ZnO buffer layer on glass substrate is very useful for improving photocatalytic properties of the present TiO₂ film. This film had a large specific surface area due to its coarse grains with about 100 nm sizes (SEM image of Fig. 5), which leads to a strong adsorption ability. The TiO₂ films are composed of oriented anatase TiO₂ (001) and rutile TiO₂ (100), which provide photo-activated electrons and holes in the surface without any influences from imperfections. Furthermore, the oxidization site of anatase TiO_2 is mainly on the (001) face which also improves [15], photocatalytic activities.

4. CONCLUSIONS

We prepared oriented TiO_2 films using a highly oriented ZnO (0001) buffer layer on quartz glass substrate by PLD. Films were analyzed by X-ray diffraction and ion beam channeling techniques. The TiO₂ films were composed of ~100 nm sizes grains with anatase TiO₂ (001) and rutile TiO₂ (100) oriented growth. It is found that the anatase-rutile ratio in such phase mixed films is controlled by oxygen pressure during the These TiO₂ films have deposition. photocatalytic activity for decomposition of organic dyes higher than that of TiO₂ films on substrate. The improvement in the glass photocatalytic activity of TiO₂ film using a ZnO buffer layer is due to increase in surface area and highly oriented crystal growth.

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