# The Impact of Rapid Thermal Annealing on ALCVD-Al<sub>2</sub>O<sub>3</sub>/Si<sub>3</sub>N<sub>4</sub>/Si(100) Stack Structures -Photoelectron Spectroscopy

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## ABSTRACT

We studied chemical bonding features and electronic defect states of dielectric stack structures consisting of 5nm-thick  $Al_2O_3$  and 1.7nm-thick  $Si_3N_4$  on Si(100) before and after rapid thermal anneal (RTA) in the temperature range of  $850 \sim 1000^{\circ}$ C for 1min in  $N_2$  or  $O_2$  ambience using X-ray photoelectron spectroscopy (XPS) and total photoelectron yield spectroscopy (PYS). The compositional intermixing at the  $Al_2O_3/Si_3N_4$  interface becomes significant with RTA higher than 900°C. For samples after RTA higher than 950°C in both  $N_2$  and  $O_2$ , Si and N atoms are incorporated throughout the  $Al_2O_3$  layer and detected even on the  $Al_2O_3$  surface. It was noted that the N content in the dielectric layer after  $O_2$ -RTA is smaller than that after  $N_2$ -RTA. This result implies the reaction of diffused N atoms with oxygen and resultant N desorption from the dielectric layer. From PYS measurements, it is found that in the energy region shallower than Si midgap and above, the defect state density of the dielectric after 950°C  $O_2$ -RTA becomes higher by a factor of 2~3 than the case after 950°C  $N_2$  RTA, which might be correlated to the N desorption.

Key words: DRAM, Al<sub>2</sub>O<sub>3</sub>/Si<sub>3</sub>N<sub>4</sub>, XPS, PYS

## 1. INTRODUCTION

Continuous scaling of dynamic (DRAM) random-access memory storage capacitors with shrinking cell area has required the use of a thinner dielectric with a higher dielectric constant (high-k) to maintain charge storage level and charge retention being below some limit on the leakage. In this regard, many studies on high-k dielectrics such as Ta<sub>2</sub>O<sub>5</sub> with a dielectric constant of ~25 were performed [1] and, currently, Ta<sub>2</sub>O<sub>5</sub> films are widely used for DRAM capacitor dielectrics. However, from the viewpoints of thermal stability and potential barrier height, Al<sub>2</sub>O<sub>3</sub>-based films are attracting much attention as the next capacitor dielectrics for further shrinkage of capacitor cells [2]. One of the major difficulties in the scaling of Ta2O5-based capacitor cells is the thickness reduction of an interfacial barrier layer compensating a low potential barrier height of Because the interfacial layer has a Ta<sub>2</sub>O<sub>5</sub>. dielectric constant much lower than Ta<sub>2</sub>O<sub>5</sub>, the reduction of the total equivalent dielectric significantly thickness is limited by the interfacial layer thickness. Also, thermal and of chemical stabilities Ta<sub>2</sub>O<sub>5</sub> against crystallization and interfacial reactions remain a significant barrier to fabricate highly reliable memory cells with a good performance in high temperature processes for transistors. On the other hand, Al<sub>2</sub>O<sub>3</sub> is showing promise to such drawbacks of overcome Ta<sub>2</sub>O<sub>5</sub>-based dielectrics because of its excellent thermal stability with silicon, high crystallization over 900°C [3], and favorable potential barrier height higher than 2eV for both electrons and holes [4, 5]. In fact, a good electrical performance of trench capacitors with an  $Al_2O_3$  layer formed by an atomic-layer chemical vapor deposition (ALCVD) has been reported [6, 7]. For the practical implementation of  $Al_2O_3$  to DRAM capacitors, one major concern is how to minimize interfacial reactions between  $Al_2O_3$  and a base material during CVD and post-deposition annealing at high temperature [8]. Thus, a clear insight into the interfacial reactions is definitely needed for better control of interface and film properties.

In this work, we focused on  $Al_2O_3/Si_3N_4$ stack structures formed on Si(100) and evaluated changes in chemical bonding features and electronic defect states in the dielectric stack with rapid thermal annealing (RTA) in N<sub>2</sub> or O<sub>2</sub> ambience.

## 2. EXPERIMENTAL

After standard chemical cleaning steps of 300mm p-type Si(100) wafers with a resistivity of  $8\sim12\Omega\cdot cm$ , the Si(100) wafer surface was nitrided in NH<sub>3</sub> ambience at a temperature of 1100°C to form a uniform ~1.7nm-thick Si<sub>3</sub>N<sub>4</sub> layer. Subsequently, a ~5.0nm-thick Al<sub>2</sub>O<sub>3</sub> layer was grown on the Si<sub>3</sub>N<sub>4</sub> layer by an ALCVD technique at a temperature of 450°C, where trimethyl-aluminum (Al(CH<sub>3</sub>)<sub>3</sub>) and O<sub>3</sub> were used as precursors. The Al<sub>2</sub>O<sub>3</sub>/Si<sub>3</sub>N<sub>4</sub>/Si(100) stack

structures so prepared were annealed in the temperature range of 850~1000°C for 1min in pure  $N_2$  (N<sub>2</sub>-RTA) or dry  $O_2$  (O<sub>2</sub>-RTA). The compositional profiles and chemical bonding features of the samples were examined by x-ray photoelectron spectroscopy using (XPS) monochromatized AlK<sub>a</sub> (1486.7eV) radiation in combination with wet-chemical thinning of the dielectric layers in a dilute HF solution. The defect state densities in the stack structures were photoelectron measured bv total yield spectroscopy (PYS) [9].

#### 3. RESULTS AND DISCUSSION

The Si2p spectra taken after N2- and O2-RTA at 1000°C are compared with the Si2p spectrum taken before RTA in Fig. 1. For the Si2p signals due to Si-N bonds peaked around ~102eV, the signal intensity of the higher binding energy side is markedly increased by the N<sub>2</sub>- and O<sub>2</sub>-RTA. This result indicates the formation of Si-O bonds. It is interesting to note that in the sample after the N2-RTA, the chemically shifted Si2p peak appears at higher binding energy by ~0.2eV than the peak after the O2-RTA. Since a similar energy shift is observed for the Al2p, O1s and N1s spectra as shown in Fig. 2, it is likely that the N2-RTA causes the generation of positive charges in the stack structure or the annihilation or neutralization of negative charges possibly existing in the as-deposited Al<sub>2</sub>O<sub>3</sub> layer. In addition, the N1s signal intensity normalized by the Si2p intensity for the sample after the O2-RTA is small compared with that after the N2-RTA, in contrast to intense O1s and Al2p signals. This result implies the desorption of N atoms by the O2-RTA.

Figure 3 shows the changes in Al2p, Si2p and N1s spectra for the sample after N<sub>2</sub>-RTA at 1000°C with progressive HF-etching steps. The chemically shifted Si2p and N1s signals from the dielectric stack are decreased with progressive thinning of the Al<sub>2</sub>O<sub>3</sub> layer. This suggests that Si and N atoms in the Si<sub>3</sub>N<sub>4</sub> layer formed at 1100°C are diffused and incorporated into the Al<sub>2</sub>O<sub>3</sub> layer. Also, considering the fact that in the early stages of Al<sub>2</sub>O<sub>3</sub> thinning (within 30s in etch time), slight decreases in the chemically shifted Si2p and N1s signals

are observable, indicating that some Si and N atoms are diffused close to the surface. In Fig. 4, for the samples before and after N2- and O2-RTA at different temperatures, the integrated intensities of chemically shifted Si2p and N1s signals are plotted as a function of the integrated intensity of A12p signals, where all these signals were normalized by the Si2p signal intensity The results obtained for the cases of from Si(100). 1000°C and 950°C RTA are very similar and irrespective of annealing ambience. Even in the early stages of wet-chemical etching, Si2p and N1s signals are decreased with a reduction in the Al2p signals and disappear slightly earlier than Al2p signals. This indicates that thin Si<sub>3</sub>N<sub>4</sub> layer can not survive by RTA over 950°C and Si and N atoms diffuse near the surface, that is, almost complete compositional intermixing occurs in the stack structure. In contrast, for the case after 850°C RTA, the Si2p signals are reduced only by 20% from the initial intensity when no Al2p signal is detectable, suggesting that the Si<sub>3</sub>N<sub>4</sub> layer remains after the RTA although some Si atoms diffuse into the Al<sub>2</sub>O<sub>3</sub>



Fig. 1. Si2p spectra for as-prepared Al<sub>2</sub>O<sub>3</sub>(5.0nm) /Si<sub>3</sub>N<sub>4</sub>(1.7nm)/Si(100) (solid line) and after being annealed in N<sub>2</sub> ambience (closed dots) and in O<sub>2</sub> ambience (open dots) at 1000°C. The photoelectron take-off angle was set at 90°.



Fig. 2. Al2p (a), Ols (b) and Nls (c) spectra for Al<sub>2</sub>O<sub>3</sub>(5.0nm)/Si<sub>3</sub>N<sub>4</sub>(1.7nm)/Si(100) after being annealed in N<sub>2</sub> ambience (closed dots) and in O<sub>2</sub> ambience (open dots) at 1000°C. The binding energy was calibrated by the Si2p3/2 peak at 99.3eV for the Si(100) substrate and the photoelectron intensity was normalized by the peak intensity of Si2p signals from the Si(100) substrate.



Fig. 3. Al2p (a), Si2p (b) and N1s (c) spectra taken at each thinning step of the dielecric stack in a diluted HF solution for the sample annealed in N<sub>2</sub> ambience at 1000°C. The binding energy was calibrated by the Si2p3/2 peak at 99.3eV for the Si(100) substrate and the photoelectron intensity was normalized by the peak intensity of Si2p signals from the Si(100) substrate.



Fig. 4. Integrated intensities of chemically shifted Si2p and N1s signals as a function of integrated Al2p signal intensity for as-deposited Al2O<sub>3</sub>/Si<sub>3</sub>N<sub>4</sub>/Si and annealed in N<sub>2</sub> (a) and O<sub>2</sub> (b) ambience at different temperatures.

layer. Also, since the N1s intensity becomes markedly smaller than the simulated result for the case of ideal Al<sub>2</sub>O<sub>3</sub>/Si<sub>3</sub>N<sub>4</sub> stack as the Al<sub>2</sub>O<sub>3</sub> thinning proceeds close to the interfacial region, it is likely that a compositional mixing between Al<sub>2</sub>O<sub>3</sub> and Si<sub>3</sub>N<sub>4</sub> occurred in an atomic layer scale. Considering there was no difference in N1s intensity between the cases before and after the 850°C RTA, the interfacial reaction in the early stages of Al<sub>2</sub>O<sub>3</sub> ALCVD on Si<sub>3</sub>N<sub>4</sub> might be responsible for such a slight intermixing. In the simulation, an increase in the N1s intensity with decreasing Al2p intensity is attributed to the fact that the escape depth of the N1s photoelectrons in the Al<sub>2</sub>O<sub>3</sub> film is smaller than that of the Si2p photoelectrons by a factor of 0.83. It is also found that there is no significant difference between the results obtained after N2- and O2-RTA except that O2-RTA at 950 and 1000°C causes a slight reduction of N atoms in the dielectric stack.

To examine the influence of the composition

intermixing gives on the electronic defect state density in the stack structures, photoelectron yields from the samples before and after RTA were measured as a function of incident photon energy as shown in Fig. 5. For incident photons with energies higher than ~5.2eV, the electron emission from the Si valence band becomes significant as indicated in the region colored in gray. By Al<sub>2</sub>O<sub>3</sub> ALCVD on Si<sub>3</sub>N<sub>4</sub>/Si(100), the photoelectron yield in the energy region below ~5.2eV is remarkably increased, which implies existence of filled defect states in the Al<sub>2</sub>O<sub>3</sub> layer and/or defect generation at the interface between Al2O3 and Si3N4. No significant change in the yield is caused by RTA at 850°C. When the RTA temperature rises up to 950°C, the photoelectron yield is decreased by more than one order of magnitude; namely, the filled defect states in the as-prepared Al<sub>2</sub>O<sub>3</sub>/Si<sub>3</sub>N<sub>4</sub> stack structure are reduced well. Note that the photoelectron yield after O2-RTA at 950°C is 2~3 times higher than that of  $N_2$ -RTA (Fig. 5(b)).



Fig. 5. Total photoelectron yield spectra for as-prepared  $Al_2O_3(5.0nm)/Si_3N_4(1.7nm)/p-Si(100)$  and after annealed in  $N_2$  ambience (a) and comparison with the yield spectra for the cases annealed in  $O_2$  ambience. The yield spectrum of as-nitrided Si(100) is also shown as a reference.  $E_v$  denotes the Si valence band top measured from the vacuum level.

Such a tendency of a higher yield being obtained by O2-RTA than N2-RTA, is also found in the cases at 850 and 900°C RTA. Based on the result obtained by XPS measurements (see Fig. 4), the desorption of N atoms from the dielectric stack caused by O2-RTA is likely to impede the defect annihilation in the dielectric stack structure or may create vacancies acting as electronic defects. From the first derivative of the PYS spectra with respect to photon energy, we crudely estimated the defect state density in the dielectric stack structure. where the density of state in the Si valence band and the photoelectron escape depth were taken into account. Details on converting the measured PYS spectrum to the energy distribution of filled defect states are described in Ref. 9. The filled defect state density in as-deposited and O<sub>2</sub>-RTA at 850°C is of the order of 10<sup>19</sup> eV<sup>-1</sup>cm<sup>-3</sup> at an energy near the Si midgap, and by N2-RTA at 950°C and higher, it is decreased efficiently and to  $-3x10^{17}$ eV<sup>-1</sup>cm<sup>-3</sup> at an energy near the Si midgap.

#### 4. CONCLUSIONS

We have studied chemical bonding features and defect densities of Al<sub>2</sub>O<sub>3</sub>/Si<sub>3</sub>N<sub>4</sub>/Si(100) stack structures before and after annealing in the temperature range of 850~1000°C in  $N_2$  and  $O_2$ ambience using photoelectron spectroscopy. In the RTA process higher than 900°C, Si and N atoms in the ultrathin Si<sub>3</sub>N<sub>4</sub> layer formed in NH<sub>3</sub> gas ambience at 1100°C diffuse significantly into the  $Al_2O_3$  layer. The RTA at 900°C or higher is very effective in decreasing the defect density in the stack structures. The residual filled defect density in RTA in the O<sub>2</sub> ambience tends to be higher than that in N<sub>2</sub> ambience. The N desorption from the dielectric layer resulting from the reaction of diffused N atoms in the Al<sub>2</sub>O<sub>3</sub> layer with oxygen may involve higher residual defects in O2-RTA than in N2-RTA.

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