

An Unfavorable Effect of Nitrogen Incorporation on Reduction in the Oxygen Vacancy Formation Energy in Hf-based High- k Gate Oxides

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An effect of N incorporation into Hf-based high- k gate dielectrics has been studied in terms of formation of oxygen vacancies (V_{O} s). The energy required to form a neutral vacancy (V_{O}^0) is calculated from the first-principles total energy calculations. Our computational results clearly showed that the V_{O}^0 formation energy is about 0.4 eV smaller in the vicinity of N atoms than that in pure HfO_2 , indicating N incorporation accelerates V_{O}^0 formation in HfO_xN_y . The additional V_{O}^0 s formed in HfO_xN_y , in turn, induce generation of fixed positive charge in the high- k bulk of MISFETs due to electron transfers from the occupied V_{O} levels to the low-lying Fermi level of the metal gate. This computational result reasonably explains a recent experimental result showing a negative shift of the flat band voltage upon N incorporation into TiN/ HfO_2 MISFET. Moreover, a possible solution to this problem is suggested.

Key words: high- k , HfO_2 , nitrogen incorporation, oxygen vacancy

1. INTRODUCTION

Oxygen vacancies (V_{O} s) in Hf-based high- k dielectrics are a nuisance because they significantly degrade the reliability of CMOS devices. For example, the gate leakage current through HfO_2 is considered to be mediated by electron hopping through V_{O} related gap states^{1,2} located about 0.4 eV above the bottom of the Si conduction band.³ The V_{O} s are also known to be electron and hole trap sites,^{4,5} which induce bias temperature instability.²

N incorporation is known to be effective for reducing both leakage current^{6,9} and electron charge traps.^{10,11} Recently, we have found that N atoms couple favorably with V_{O} s in HfO_xN_y and deactivate the V_{O} related gap states by extracting electrons from them, which leads to a reduction in the gate leakage current.¹²

The same effect has been observed in HfSiON .¹³ Moreover, our subsequent work revealed that electron charge traps at V_{O} s are also suppressed by N incorporation.¹⁴ There are many other beneficial effects of N incorporation such as increasing the dielectric constant⁶ and inhibiting crystallization.^{6,7,15,16} Therefore, N incorporation into high- k dielectrics has become a standard technique in fabricating good quality gate insulators. Very recently, however, one adverse effect of N incorporation has been reported,¹⁷ i.e. N incorporation into TiN/ HfO_2 MISFET causes a negative shift of the flat band voltage indicating the existence of fixed positive charge in the high- k gate insulator. In the present study, we focused on the possibility that the positive charge is generated inside the high- k bulk due to N incorporation. Our suggested scenario is schematically illustrated in Fig. 1. If the energy required to form neutral V_{O}^0 s in the vicinity of the N atoms is relatively small, the V_{O}^0 s are selectively formed around

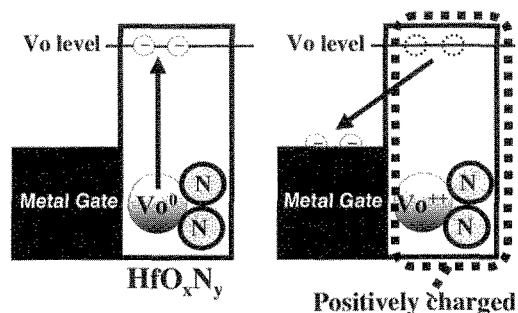


Fig. 1. Schematic illustration of suggested scenario for the generation of fixed positive charge induced by N incorporation.

the N atoms. Thus, the number of occupied V_{O} levels in HfO_xN_y is increased. This, in turn, causes the generation of positive charge because of the electron transfer from the V_{O} levels to the low-lying Fermi level of the metal gate. This scenario is based on the assumption that “the V_{O}^0 s are favorably formed around N atoms”. Therefore, the present work is devoted to confirming this assumption in order to justify the above scenario. For this purpose, we have compared the relative stabilities of the V_{O} formation energies in HfO_2 and HfO_xN_y using the first-principles total energy calculations. Our computational results clearly show that the V_{O} formation energy is certainly reduced by 0.4 eV in the vicinity of N atoms. This indicates that N incorporation accelerates V_{O} formation, which, in turn, causes the generation of fixed positive charge. Moreover, our theoretical results suggest a possible solution to this problem.

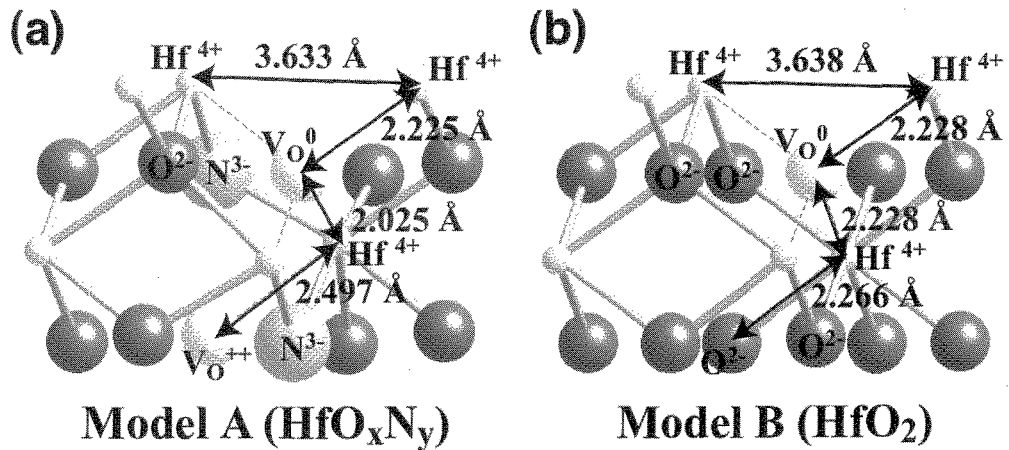


Fig. 2. Relaxed structures in the vicinity of a Vo^0 for the models of HfO_xN_y and HfO_2 . (a) In model A, two N^{3-} ions and one Vo^{++} are introduced into the unit cell to satisfy the charge neutrality of the system. On top of that, a Vo^0 is also introduced to calculate the Vo^0 formation energy. (b) In model B, one oxygen atom is removed from the unit cell of the cubic HfO_2 to model the formation of a Vo^0 formation in HfO_2 .

2. CALCULATIONS

Our theoretical calculations are based on the first-principles pseudopotential¹⁸ approach within the framework of a generalized gradient approximation (GGA).^{19,20} The models employed for HfO_xN_y and HfO_2 are based on cubic HfO_2 including 96 atoms per unit cell. The cutoff energy of the plane-wave basis set is 36 Ry. The Brillouin zone (BZ) was sampled at the Γ point. The convergence of the calculation parameters were checked by increasing the cutoff from 36 to 49 Ry, and extending the BZ sampling to eight k points beyond the Γ point. The geometries of the models employed were fully relaxed for all atoms until the residual forces were less than 4 mRy/Å. In Fig. 2, the relaxed structures around Vo s are depicted. Here, the places of the Vo s are represented by the locations of Vo s in the initial structures before geometrical optimization. In model A, which represents HfO_xN_y , two O atoms are replaced by N atoms and one Vo^{++} and one Vo^0 are introduced into the unit cell (Fig. 2 (a)). The composition of one Vo^{++} and two N^{3-} ions is convenient to retain the charge neutrality of the system. Thus, the formation energy of a neutral Vo^0 is correctly calculated. Here, N atoms are placed at the nearest neighbor oxygen sites to the Vo s because the N atoms couple favorably with Vo s.¹² In model B, which represents HfO_2 , one O atom is removed from the unit cell of the cubic HfO_2 to introduce a neutral Vo^0 (Fig. 2 (b)). We found that most of the Hf-Hf and Hf- Vo^0 distances are not very much different between models A and B except for one Hf- Vo^0 distance; 2.025Å in HfO_xN_y , and 2.228Å in HfO_2 . The difference in the Hf- Vo^0 distances is attributed to the relaxation of Hf^{4+} ions around Vo^{++} in model A.¹² We consider that this slight difference between models A and B does not significantly affect the comparison of Vo^0 formation energies between these models. We have also calculated the total energies of these models without the Vo^0 . Finally, the energies required to form a Vo^0 per unit cell, i.e., the Vo^0

formation energies: $F(\text{Vo})$ in HfO_xN_y and HfO_2 are given respectively by,

$$F_{\text{HFON}}(\text{Vo}) = \left[E_{\text{HFON}+2\text{Vo}}^0 + \frac{E_{\text{O}_2}}{2} \right] - E_{\text{HFON}+\text{Vo}}^0, \quad (1)$$

$$F_{\text{HFO}_2}(\text{Vo}) = \left[E_{\text{HFO}_2+\text{Vo}}^0 + \frac{E_{\text{O}_2}}{2} \right] - E_{\text{HFO}_2}^0, \quad (2)$$

where $E_{\text{HFON}+2\text{Vo}}^0$ is the total energy of the neutral state of the $\text{HfO}_x\text{N}_y + 2\text{Vo}$ system. E_{O_2} is the total energy of an oxygen molecule calculated with a unit cell the same size of as in models A and B. Our computational results obtained from Eqs. (1) and (2) clearly show that the Vo formation energy of model A (HfO_xN_y) is 0.4 eV smaller than that of model B (HfO_2). This indicates that Vo^0 formation is augmented around N atoms. This result justifies the assumption used for our scenario addressed in Fig. 1.

3. PHYSICAL REASONS

A qualitative understanding of this effect due to N incorporation can be derived from the relatively strong repulsive Coulomb interaction between N^{3-} and O^{2-} ions as schematically illustrated in Fig. 3. In the case of HfO_xN_y , O^{2-} ions located near to N atoms are subject to strong repulsive Coulomb potentials from N^{3-} ions. Thus, the O^{2-} ions around N^{3-} ions are relatively unstable. The repulsive interactions are essentially reduced by the removal of a neutral O atom from around the N^{3-} ion because of the redistribution of electrons (Fig. 3 (a)). On the other hand, O^{2-} is already stable in HfO_2 and a relatively large amount of energy is required to remove a neutral O atom compared to the case in HfO_xN_y . The small redistribution of electrons in the Vo^0 also contributes to the larger Vo^0 formation energy in HfO_2 (Fig.3 (b)). After formation of the neutral Vo^0 , two electrons occupy the Vo levels that are usually located

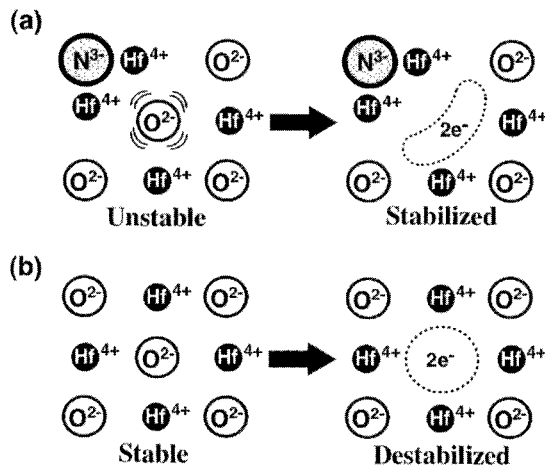


Fig. 3. Schematic illustrations of repulsive Coulomb interactions between negative ions and electrons in (a) HfO_xN_y and (b) HfO_2 before and after Vo^0 formation.

above the metal Fermi level. Thus, these electrons can transfer to the metal gate of the MISFET. Consequently, positive charge is generated in the high- k dielectrics as illustrated in Fig. 1. This is a similar mechanism to the Vo -induced Fermi level pinning in p+poly Si gates.^{21,22} This effect is undesirable because the additional Vo^0 s formation causes the generation of fixed positive charge in MISFETs that leads to negative shift of the flat band voltage and possibly degrades mobility. Therefore, adjustment of the process environment is required so that the additional Vo^0 formation in HfO_xN_y is energetically disadvantageous. Actually, it has been reported that an increase in the ambient O_2 partial pressure after the N incorporation process is effective for this purpose.²³

4. CONCLUSIONS

In conclusion, we have investigated the effect of N incorporation on the Vo formation energy in HfO_xN_y using the first-principles total energy calculations. It was found that the formation energy of Vo^0 s is essentially reduced by N incorporation, resulting in the generation of fixed positive charge in HfO_xN_y . Moreover, our theoretical analysis suggests that adjustment of the ambient O_2 partial pressure is necessary after N incorporation process to prevent the formation of additional Vo^0 s. In future, another possible location of the fixed positive charge, i.e., the high- k dielectric/Si-substrate interface should also be examined.

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